



Studies on physico-chemical parameters of Siddapura Lake, Madhugiri taluk, Tumkur dist, India

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Abstract

Water is one of the naturally occurring vital constituent of the ecosystem which supports all life activities. Fresh water is a source of drinking water and is getting polluted by the discharge of domestic sewage, industrial effluents and anthropogenic activities. Analysis of Physico-Chemical parameters of Siddapura lake Tumkur district was done during the period of July-2015 to December-2015. The water samples were collected on the monthly basis from three sampling sites. The analysis was carried out for the parameters like Temperature, pH, Electrical conductivity, Turbidity, Chlorides, Nitrates, Sulphates, Phosphates, and Biological Oxygen Demand (BOD).

Keywords: siddapura lake, pollution, physico-chemical parameters

1. Introduction

Water is one of the most important natural resource needed for the existence and development of life on earth. The aquatic system is disturbed by the contamination of water [1]. The increased human population, industrialization, use of fertilizers and manmade activities are polluting the water [2]. The quality of water is altered by the change in the physico-

chemical properties of water and affects the aquatic environment and thus causing imbalance in the ecosystem [3]. Many health hazards including water borne diseases are linked with the unsafe drinking of water. Hence, it is very much essential to check the water quality regularly [4]. The present study was carried out to assess the Physico-chemical parameters of Siddapura lake, Tumkur (Dist) India.

2. Study area



Fig 1

Siddapura lake is one of the important lakes in Tumkur District. Lake is geographically located at latitude $13^{\circ} 40'0''$ and longitude $13^{\circ} 40'0''$. It is located in between Siddapura village

and Madhugiri town on Sira road. The tank was mainly constructed for the purpose of irrigation. Now a days it is used for various purposes like agriculture, fish culture, drinking and

partially for domestic purposes.

3. Materials and Methods

Surface water samples from Siddapura lake were collected from the selected sampling sites in the morning between 7 AM to 9AM in two liters of polythene cans regularly for every month. Three sampling sites were selected which completely cover the whole area of the lake. The investigation was carried out to study the physico-chemical parameters of water during July- 2015 to December-2015. The temperature was recorded

4. Results and Discussion

Table 1: Average values of physico-chemical parameters compared with BIS standards

Month	Temp. (°C)	pH	Turbidity NTU	Electrical Conductivity µs/cm	Chloride Mg/L	Sulphate Mg/L	Phosphate Mg/L	Nitrate Mg/L	BOD Mg/L
BIS Standards	40	6.5-8.5	5	-	250	200	-	45	5
July-2015	25.9	7.4	2.9	203	24	11	0.9	0.54	4
Aug-2015	26.2	7.2	2.8	180	22	2	0.89	0.44	7
Sep-2015	26.1	7.4	2.7	176	22	2	0.92	0.32	10
Oct-2015	27.1	7.4	3.4	179	25	2	0.83	0.31	15
Nov-2015	26.3	7.4	3.6	180	26	3	0.94	0.34	9
Dec-2015	25.1	8.6	2.7	194	20	2	0.91	0.35	4

5. Temperature

Temperature is one of the important parameter which affects the various activities of the aquatic organisms like respiration, metabolism etc. It depends on the environmental factors like rainfall, humidity, weather, wind etc. The rate of all chemical reactions is regulated by the water temperature which in turn affects the growth and reproduction of fishes [9]. In the study period, water temperature ranged from 25.1°C to 27.1 °C. The maximum water temperature (27.1°C) was recorded in October-2015 and minimum water temperature (25.1°C) was recorded during December-2015.

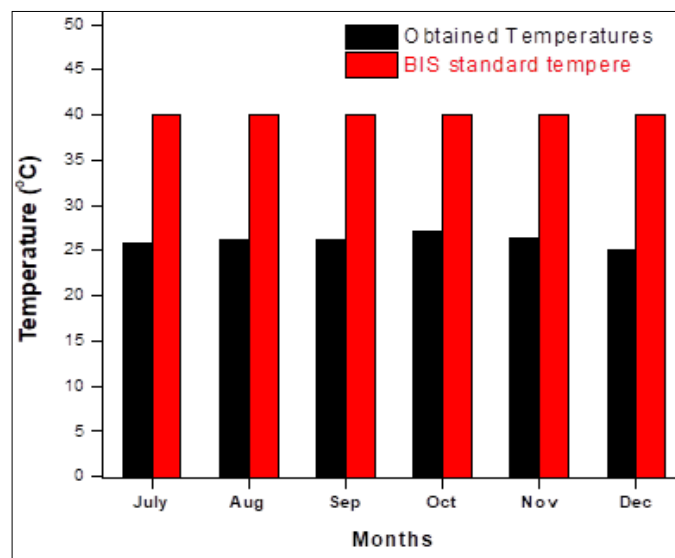


Fig 2: Comparison of Temperature values with BIS standards

6. Turbidity

In a water sample, turbidity measures the amount of light scattered by the particles. The colloidal solution is formed by

at the spot using Thermometer. In addition to temperature, other physico-chemical parameters such as pH-Electrometric method, Turbidity-Nephelometric method, Electrical conductivity- Electrometric method, Chloride-Aargentometric method, Nitrate-Phenol di-sulphonic acid method, Sulphate-Turbidimetric method, Biological Oxygen Demand (BOD)-Winkler’s Iodometric method were estimated for the collected samples in the laboratory using standard methods [5]. The results were compared with standard permissible limit by BIS [6].

the aggregation of suspended particles. The extremely minute particles occupy the surface bottom and minute particles remain as suspended which forms turbidity. Turbidity of water ranges from 2.7 to 3.6NTU. The highest value of 3.6NTU was observed in November-2014 and lowest value of 2.7 NTU was observed in September and December- 2015.

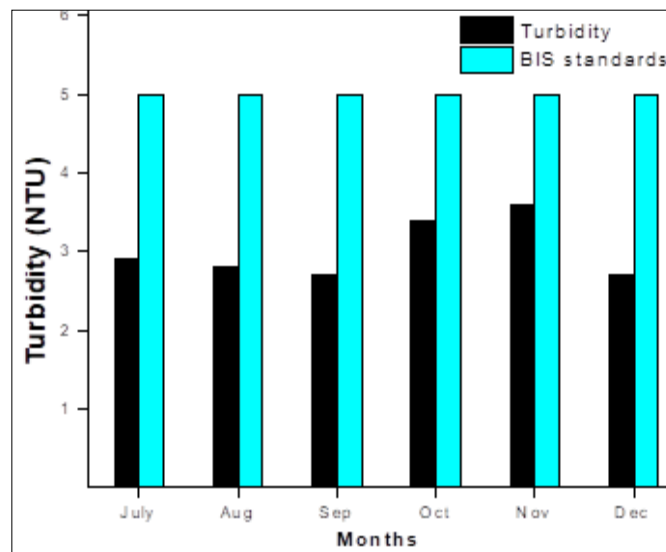


Fig 3: Comparison of Turbidity values with BIS standards

7. pH

pH is defined as the intensity of acidic or basic character of a solution at given temperature. It is one of the important physico-chemical factors for estimating the water quality. The acidic and alkaline nature of water is determined by the pH. The pH of water is regulated by carbon dioxide and bicarbonates. It is affected by the factors like Temperature, salinity.

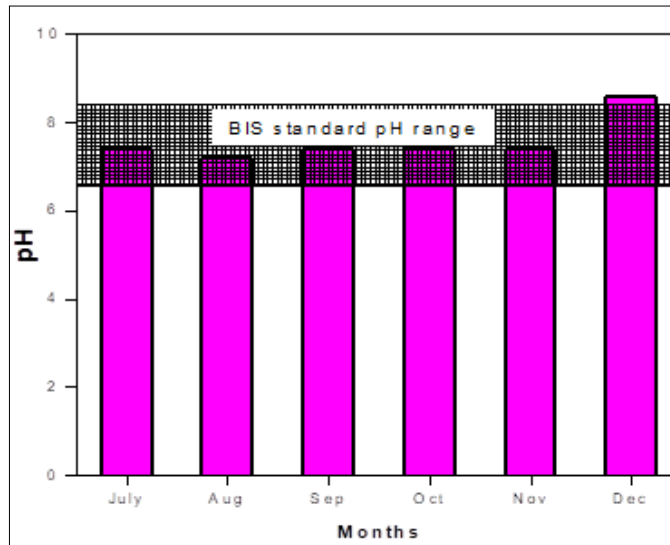


Fig 4: Comparison of pH values with BIS standards

In the current study the pH was alkaline and found in the range between 7.2 to 8.6. The maximum value of pH (7.2) was recorded in August-2015 and minimum value of pH (8.6) was recorded in December-2015

8. Electrical conductivity

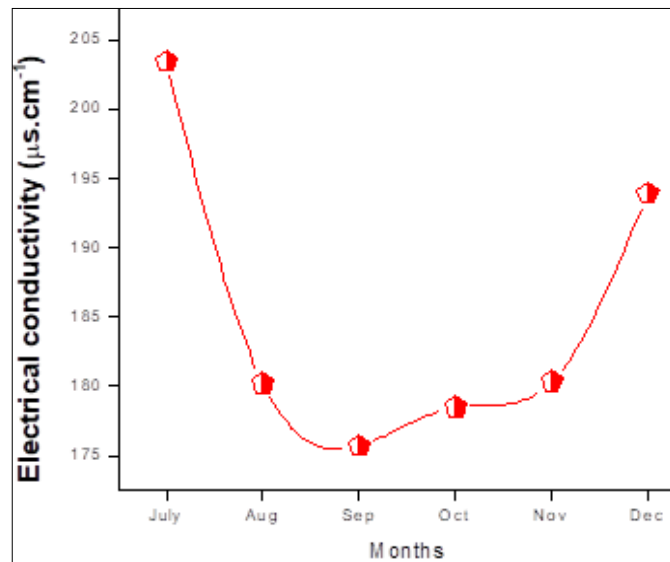


Fig 5: Electrical conductivity of water samples

Electrical conductivity (EC) is the ability of aqueous solution to carry electrical current. It depends on the presence of ions and their concentrations, temperature. The increase in EC of water indicates the degree of pollution. The water quality can be checked by controlling the conductivity of water. The Electrical conductivity in the present study was ranged from 176 mg/L to 203 mg/L. The maximum value of EC (203 mg/L) was recorded in July-2015 and minimum value of EC (176 mg/L) was recorded in September-2015.

9. Sulphates

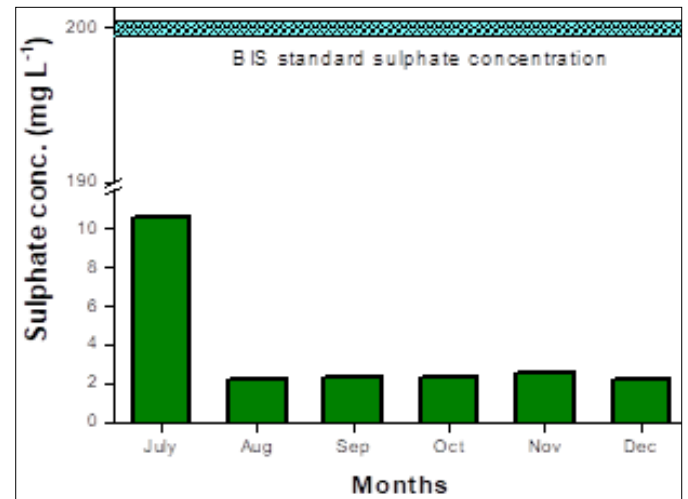


Fig 6: Comparison of Sulphate values with BIS standards

Sulphur is added to the water through runoff water containing fertilizers, organic and mineral sulphur compounds and by the dissolution of minerals from soil and rocks. High amount of sulphate induces bitter taste to water. The values of Sulphates in the present study ranged from 2 mg/L to 11 mg/L. The maximum value (11 mg/L) was recorded in July-2014 and minimum value (2 mg/L) was recorded in August- 2015 and December-2015.

10. Phosphates

Phosphate is a fundamental element in the metabolic reactions of plants and animals. It is essentially found in bones and teeth. It controls the growth of algae and primary productivity. The increased phosphates in water leads to eutrophication and results in the excessive algal growth called Algal bloom.

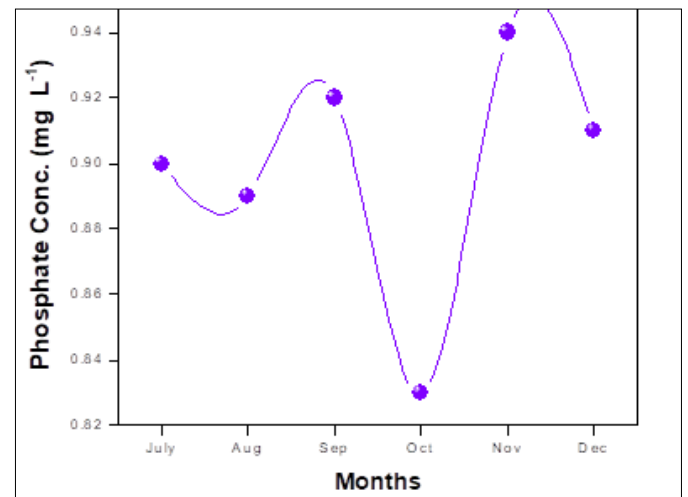


Fig 7: Comparison of Phosphate values with BIS standards

The values of Phosphates fluctuate from 0.83 mg/L to 0.94 mg/L. Maximum value of Phosphates (0.94 mg/L) was

recorded in November-2015 and minimum value of Phosphates (0.83 mg/L) was recorded in October-2015.

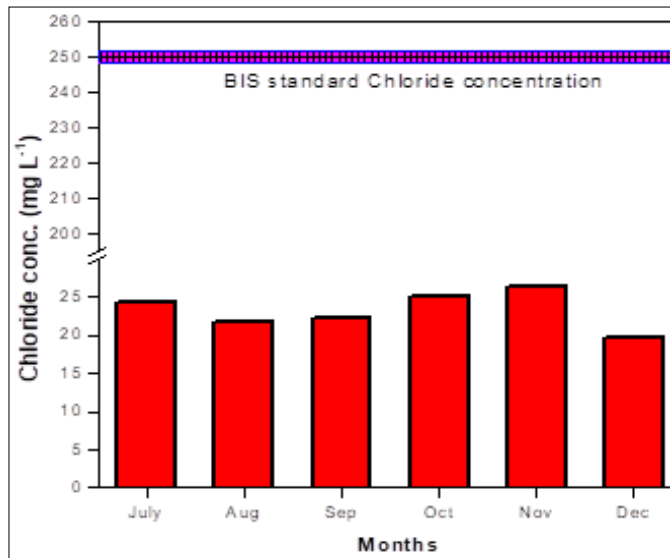


Fig 8: Comparison of Chloride values with BIS standards

11. Chlorides

Chlorides play a very important role in determining the quality of water. The chloride concentration is higher in organic wastes.

The higher chloride concentration indicates the increased level of water pollution from the organic wastes [8]. The values of Chlorides fluctuate from 20 mg/L to 26 mg/L. Maximum value of Chlorides (26 mg/L) was recorded in November-2015 and minimum value of Chlorides (20 mg/L) was recorded in December-2015.

12. Nitrates

Nitrate is a very good parameter to decide the organic pollution of water and indicate the highest oxidized form of nitrogen [7]. In the present study, the values of Nitrates ranged from

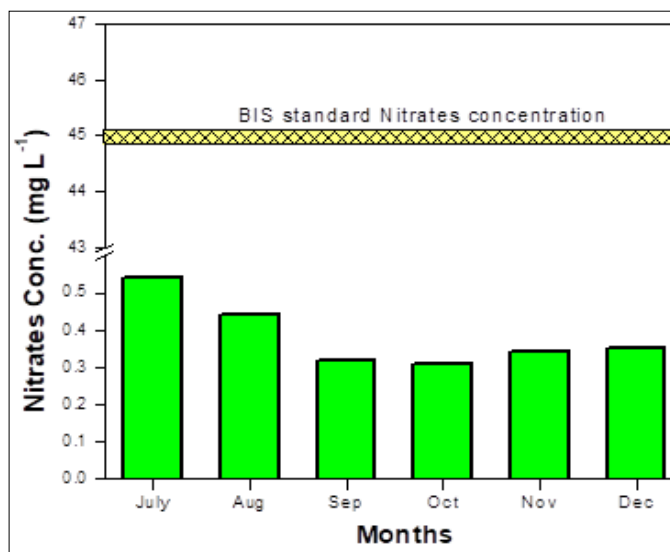


Fig 9: Comparison of Nitrate values with BIS standards

0.31mg/L to 0.54mg/L. The maximum value of Nitrates (0.54mg/L) was observed in July-2015 and minimum value of Nitrates (0.31mg/L) was observed in October-2015.

13. BOD

BOD is the quantity of dissolved oxygen required for the organic matter under test solution. It indicates the magnitude of water pollution by oxidizable organic matter.

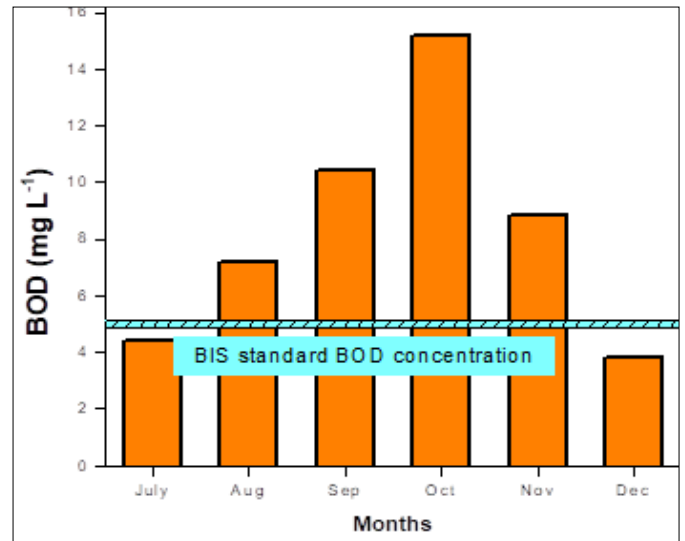


Fig 10: Comparison of BOD values with BIS standards

The main sources of organic pollution are domestic sewage, agricultural runoff and industrial effluents. The values of BOD ranges from 4 mg/L to 15 mg/L. The maximum value of BOD (15 mg/L) was found in September-2015 and the minimum value of BOD (4 mg/L) was found in December -2015.

14. Conclusion

From the present study, it is concluded that pH, Temperature, Turbidity, Electrical conductivity, Sulphate, Phosphate, Chloride and Nitrate are within the Permissible limits of BIS standard except BOD which was beyond the permissible limit. During the study period, in the month of August, September, October and November- 2015, the BOD was recorded beyond the permissible limit. This may be due to the reduction in the level of water and increased human activities in the tank. Hence, the water can be used for agriculture and in this period, the tank water is used for drinking purpose only after the treatment.

15. References

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